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# Phytochemical Study of *Eryngium triquetrum*: Isolation of Polyacetylenes and Lignans

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## Key words

- *Eryngium triquetrum*
- Apiaceae
- polyacetylenes
- lignans
- NMR

## Abstract

Phytochemical investigation of the ethyl acetate extract from the aerial parts of *Eryngium triquetrum* Vahl resulted in the isolation of new polyacetylenes, triquetridiol (**6**) and *trans*-epoxy-triquetrol (**7a/7b**, diastereomeric mixture), and the lignan demethoxy carolignan Z (**8a/8b**, *erythro/threo* pair), together with a series of re-

lated known metabolites. Additionally, some already reported phenolic and flavonoid compounds were also identified in the extract. Structural elucidation of the new compounds was made by spectroscopic analysis, mainly NMR and mass spectrometry. To the best of our knowledge, this is the first report of polyacetylenes and lignans from *E. triquetrum*.

## Introduction

The genus *Eryngium* belongs to the family Apiaceae (Umbelliferae) and contains about 317 species [1]. Some of them, such as *Eryngium campestre* L., *Eryngium creticum*, *Eryngium kotschyi*, *Eryngium maritimum* L., and *Eryngium trisetum*, are used as a folk remedy for the treatment of various anti-inflammatory disorders. Some other species, for instance, *Eryngium falcatum*, are known for their antinociceptive activity. Secondary metabolites isolated from plants belonging to this genus have displayed important biological activities, including antitumor, antibacterial, antimicrobial, antifungal, phototoxic, and other chemical and medicinal properties [2, 3]. The phytochemical constituents of the *Eryngium* genus (23 studied species) including terpenoids, polyacetylenes, triterpenoid saponins, steroids, and phenolics such as flavonoids and coumarins, have been recently reviewed in a comprehensive article [4].

*Eryngium triquetrum* Vahl is an endemic North Africa plant widely distributed in all parts of Algeria along with other *Eryngium* species such as *Eryngium dichotomum* Desf., *Eryngium barrelieri* Boiss., *Eryngium ilicifolium* Lam., *E. maritimum* L., *E. campestre* L., and *Eryngium tricuspdatum* [5]. *E. triquetrum* grows particularly well in rocky pastures and it is known as “*Choukzerka*” by local people. To the best of our knowledge, only a single

paper has recently been published [6] that describes the chemistry of the essential oils and flavonoids of this plant.

Here we describe the chemical investigation of the ethyl acetate extract of the aerial part of *E. triquetrum* Vahl collected in Merouana (Algeria), which showed a remarkable and complex secondary metabolite pattern, including molecules of different structural classes. This study led to the finding of 20 compounds – polyacetylenes, lignans, and phenolic metabolites – including two new C<sub>17</sub> polyacetylenes, triquetridiol (**6**) and *trans*-epoxy-triquetrol (**7a/7b**, diastereomeric mixture), and the inseparable *erythro/threo* pair of unprecedented demethoxy carolignan Z (**8a/8b**).

## Results and Discussion

The dried aerial part of *E. triquetrum* was exhaustively extracted with an hydroalcoholic solution and the resulting aqueous residue was treated with organic solvents of increasing polarity (see Materials and Methods for details). The ethyl acetate extract (3 g) of *E. triquetrum* was taken into consideration for chemical analysis, revealing the presence of a rich metabolite pattern. The extract was first subjected to Sephadex LH-20 in CHCl<sub>3</sub>/MeOH, 1 : 1, to give 11 main fractions. Preliminary NMR analysis of fractions 3 and 4 showed the

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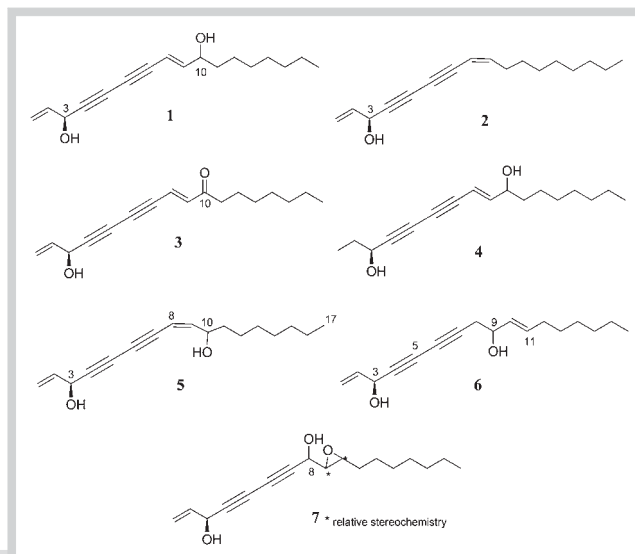
presence of signals attributable to polyacetylenes and lignans, whereas fraction 7 contained mainly flavonoids. Purification of these three fractions by SiO<sub>2</sub> chromatography, followed by semi-preparative HPLC (see Materials and Methods), afforded 20 compounds, 3 of them not previously reported, including polyacetylenic, lignan, and phenolic metabolites. The polyacetylene fraction, the less polar of the *E. triquetrum* extract, was considered first. The chemical analysis of this fraction resulted in the isolation of C<sub>17</sub> enynols (compounds 1–7; **Fig. 1**).

First of all, the main component was identified as *trans*-panaxydiol (1) (8*E*)-heptadeca-1,8-diene-4,6-diyne-3*S*,10-diol), a polyacetylene known from ginseng (*Panax ginseng*, Araliaceae) and from a number of further members of the Apiaceae and Araliaceae families [7–12]. Both (–)- and (+)- stereoisomers have been found in nature even though the absolute configuration of naturally occurring panaxydiol has not been unequivocally defined. All four possible stereoisomers [(3*R*,10*R*), (3*R*,10*S*), (3*S*,10*R*), (3*S*,10*S*)] of panaxydiol have been synthesized and their optical rotation have been measured showing that the sign of [α]<sub>D</sub> is mainly affected by the C-3 configuration (3*R* in levorotatory isomers, 3*S* in dextrorotatory isomers) [13]. The observed optical rotation ([α]<sub>D</sub> = 32.8°) for *trans*-panaxydiol (1) isolated in this work was very close to that reported for the (3*S*,10*R*)-stereoisomer ([α]<sub>D</sub> = 30.3°) [13], leading us to tentatively assign this absolute configuration. In order to further support this assignment, the modified Mosher method [14, 15] was applied to compound 1. The NMR analysis of both *S* and *R* MTPA esters of 1 substantiated the 3*S* absolute configuration (see Materials and Methods; **Fig. 2**), whereas the evaluation of Δδ values (δ<sub>S</sub> ester – δ<sub>R</sub> ester) for the protons adjacent to C-10 appeared to be quite complicated. It has been reported in the literature [11] that attempts to determine the absolute configuration of C-10 for a levorotatory panaxydiol failed due to the too small magnitude of the Δδ values observed for the adjacent protons at C-10. In our case, a careful analysis of the high-resolved proton spectra of the Mosher esters of 1 revealed that two set of signals (ratio ca. 1 : 1) were generated by the protons adjacent to C-10. As depicted in **Fig. 2**, both signals at Δδ 5.77 (d, *J* = 15.9 Hz, H-8) and Δδ 6.33 (dd, *J* = 15.9, 8.0, H-9) (line a) were each split into two distinct equivalent multiplets shifted at Δδ 5.78/5.65 and Δδ 6.26/6.18 (lines b and c), respectively, revealing that *trans*-panaxydiol isolated from *Eryngium* extract was not a single compound, but rather a mixture of C-10 epimers.

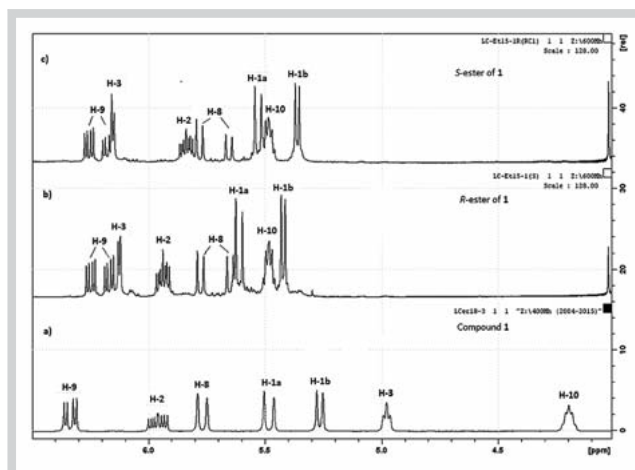
The remaining components of the mixture, including falcarinol (2), heptadeca-1,8-diene-4,6-diyne-3-ol-10-one (3), panaxjapyne B (4), *cis*-panaxydiol (5), and the unreported triquetridiol (6) and *trans*-epoxy-triquetrol (7), are described below on the basis of their polarity.

Falcarinol (2), also known as panaxynol [16, 17], was easily identified in the less polar polyacetylene fraction by NMR and MS data [18]. As for panaxydiol, both (–)- and (+)- enantiomers have been isolated from natural sources, and the absolute configuration has been established by different methods [19, 20]. The *S* absolute configuration of C-3 was indicated by the positive [α]<sub>D</sub> and confirmed by the Mosher method [19]. The NMR analysis of both *S* and *R* MTPA esters of 2 showed Δδ values consistent with the 3*S* absolute configuration (see Materials and Methods).

Heptadeca-1,8-diene-4,6-diyne-3-ol-10-one (3) [8, 21] and panaxjapyne B (4) [22–24] were isolated in a very minor amount. The 3*S* configuration for both molecules could be suggested by biogenetical considerations. The positive [α]<sub>D</sub> value of compound



**Fig. 1** Polyacetylenes isolated from *E. triquetrum*.



**Fig. 2** Selected <sup>1</sup>H NMR regions for compound 1 (line a), *R*-MTPA (line b), and *S*-MTPA (line c) esters of 1. It should be noted that two sets of H-8 and H-9 signals are present in the Mosher esters spectra (lines b and c).

4 supported this assumption, whereas, unfortunately, compound 3 degraded before the optical activity measurement.

*cis*-Panaxydiol (5) (8*Z*)-heptadeca-1,8-diene-4,6-diyne-3*S*,10-diol) has been previously reported to occur in two species of the Apiaceae family, but it has been only partially characterized, and, in particular, no NMR data have been reported [7, 8]. Analysis of <sup>1</sup>H and <sup>13</sup>C NMR resonances of 5 (**Tables 1 and 2**) revealed strong similarities with those of main co-occurring *trans*-panaxydiol (1), mainly differing in the coupling constant values of Δ<sup>8</sup> double bond protons (*J*<sub>H8–H9</sub> = 11.0 Hz in 5; *J*<sub>H8–H9</sub> = 15.9 Hz in 1, see Materials and Methods). This was consistent with a different double bond geometry, in particular, indicating the *cis*-Δ<sup>8</sup> stereochemistry in compound 5. Steric effects observed between H-8 and H-9 by selected NOE difference experiments confirmed such an assignment. Unfortunately, the easy and almost quantitative isomerization of compound 5 into 1 that was observed to occur during the chromatographic workup as well as in the NMR

**Table 1** <sup>1</sup>H NMR (400 MHz) data for compounds 5–7 in CDCl<sub>3</sub> (δ H in ppm, J in Hz). <sup>a</sup>Data divergent for minor isomers are shown in square brackets.

	5	6	7	7 (C <sub>6</sub> D <sub>6</sub> )
1	5.49, d (17.0) 5.28, d (10.2)	5.47, d (17.2) 5.25, d (10.2)	5.48, d (17.0) 5.28 d (10.1)	5.20, d (17.0) 4.86, d (10.2)
2	5.97, ddd (17.0, 10.2, 5.5)	5.95, ddd (17.2, 10.2, 5.3)	5.95, ddd (17.0, 10.1, 5.4)	5.62, ddd (17.0, 10.2, 5.4)
3	5.00, app t (5.5)	4.92, app t (5.7)	4.94, app t (5.0); OH 1.9, d (6.0)	4.47, dd (5.6, 5.4); OH 1.17 overlapped
8	5.59, d (11.0)	2.55, d ABq (6.0, 17.5)	4.66, app t (5.0); OH 2.12, d (5.0) [4.41, dd (3.8, 7.1)]; [OH 2.15, d (7.1)]	4.16, dd (3.3, 5.0); OH 1.52 d (5.0) [4.01, dd (3.7, 7.0); OH 1.59 d (7.0)]
9	6.08, dd (11.0, 8.6)	4.26, ddd (6.8, 6.0, 4.6)	3.02, m [3.00, m]	2.65, dd (3.3, 1.4) [2.68], m
10	4.64, m	5.52, dd (15.3, 6.8)	3.10, dt (2.0, 5.6) [2.98, dt (2.0, 5.6)]	2.88, br t (5.5) [2.67], m
11	1.63, m 1.51, m	5.74, dt (15.3, 6.8)	1.59, m [1.58]	1.28, m [1.27]
12	1.37, m	2.05, m	1.44, m	1.22, m
13	1.31, m	1.31, m	1.29, m	1.15, m
14	1.30, m	1.30, m	1.32, m	1.15, m
15	1.27, m	1.27, m	1.28, overlapped	1.17, m
16	1.29, m	1.29, m	1.28, overlapped	1.26, m
17	0.89, t (6.9)	0.89, t (7.1)	0.89, t (7.1)	0.89, t (6.9)

<sup>a</sup>Assignments aided by COSY, HSQC, and HMBC experiments

tube prevented an accurate measurement of the  $[\alpha]_D$  value and further stereochemical analysis. In this regard, some considerations should be made. It is reasonable to assume that *trans*-panaxydiol (**1**) isolated from *Eryngium* extract derives, in part, by the isomerization of co-occurring *cis*-panaxydiol (**5**). This implies that both isomers have the same 3S absolute configuration, as inferred by analysis of the Mosher derivatives of **1** (see above), while the stereochemistry at C-10 remains undetermined. Similar to the *trans*-isomer, *cis*-panaxydiol could be a C-10 epimeric mixture even though the possibility that the two naturally occurring isomers **1** and **5** display opposite configuration at C-10 should not be ruled out. In such a case, the transformation of **5** into **1** should explain the existence in the extract of C-10 epimers of **1**.

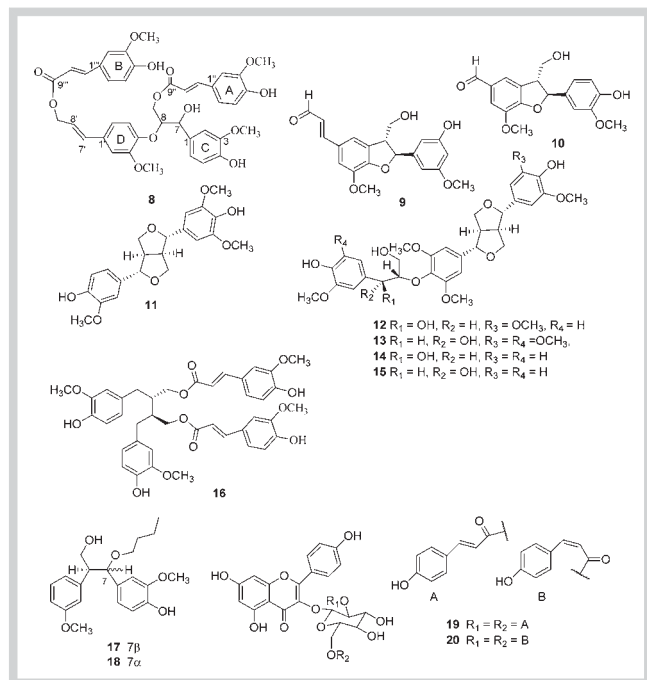
Compound **6** was obtained as a yellowish oil and its molecular formula was assigned as C<sub>17</sub>H<sub>24</sub>O<sub>2</sub> from the sodiated molecular peak at *m/z* 283.1665 [M + Na]<sup>+</sup> observed in the HRESIMS spectrum. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of this compound displayed signals very similar to those observed for known co-occurring polyacetylenes **1–5**, in particular, indicating the presence of the same partial structure C-1/C-7 (see Tables 1 and 2). The <sup>1</sup>H NMR spectrum also showed multiplets that were consistent with the presence of a *trans* double bond [ $\delta_H$  5.52 (1H, dd, *J* = 15.3, 6.8 Hz, H-10) and  $\delta_H$  5.74 (1H, dt, *J* = 15.3, 6.8 Hz, H-11)], a carbinolic methine [ $\delta_H$  4.26 (1H, ddd, *J* = 6.8, 6.0, 4.6 Hz, H-9)], a downfield shifted methylene linked to a quaternary carbon [ $\delta_H$  2.55 (2H, d ABq, *J* = 6.0, 17.5 Hz, H<sub>2</sub>-8)], an allylic methylene [ $\delta_H$  2.05, (2H, m, H<sub>2</sub>-12)], and an aliphatic chain with a terminal methyl group [ $\delta_H$  0.89 (3H, t, *J* = 7.1 Hz, H<sub>3</sub>-17)]. Analysis of the <sup>1</sup>H-<sup>1</sup>H COSY experiment led us to define the proton sequence from H-8 to terminal H<sub>3</sub>-17, whereas the heterocorrelations observed in the HSQC and HMBC spectra aided us in defining the remaining part of the molecule and to complete the structure. In particular, the H-1/H-3 and H-8/H-17 spin systems were connected through a conjugated diyne moiety by diagnostic HMBC correlations observed between H<sub>2</sub>-8 ( $\delta_H$  2.55) and both C-7 ( $\delta_C$  78.2) and C-6 ( $\delta_C$  66.5) as well as H-3 ( $\delta_H$  4.92) and C-4 ( $\delta_C$  81.5). Thus, the structure was suggested to be (10*E*)-heptadeca-1,10-diene-4,6-diyne-3,9-diol (**6**), to which the trivial name triquetridiol was given. The abso-

**Table 2** <sup>13</sup>C NMR (75 and 150 MHz) data for compounds 5–7 in CDCl<sub>3</sub> (δ C in ppm). <sup>a</sup>Data divergent for minor isomers are shown in square brackets.

	5	6	7	7 (C <sub>6</sub> D <sub>6</sub> )
1	117.4, CH <sub>2</sub>	117.2, CH <sub>2</sub>	117.2, CH <sub>2</sub>	116.1, CH <sub>2</sub>
2	135.8, CH	135.8, CH	135.8, CH	136.0, CH
3	63.7, CH	63.5, CH	63.3, CH	63.0, CH
4	81.1, C	81.5, C	nd	79.2, C
5	70.9, C	Nd	nd	70.3, C
6	nd	66.5, C	nd	nd
7	75.2, C	78.2, C	77.2, C	nd
8	108.4, CH	28.6, CH <sub>2</sub>	61.3, CH [62.3]	61.7, CH [62.5]
9	150.0, CH	70.8, CH	58.8, CH [59.3]	59.0, CH [59.6]
10	70.4, CH	130.4, CH	56.0, CH [56.1]	55.7, CH [55.5]
11	36.6, CH <sub>2</sub>	133.9, CH	31.2, CH <sub>2</sub>	31.0, CH <sub>2</sub>
12	25.0, CH <sub>2</sub>	32.1, CH <sub>2</sub>	25.7, CH <sub>2</sub>	25.8, CH <sub>2</sub>
13	29.2, CH <sub>2</sub>	28.9, CH <sub>2</sub>	28.5, CH <sub>2</sub>	29.3, CH <sub>2</sub>
14	29.5, CH <sub>2</sub>	29.0, CH <sub>2</sub>	29.0, CH <sub>2</sub>	29.3, CH <sub>2</sub>
15	31.8, CH <sub>2</sub>	31.7, CH <sub>2</sub>	31.7, CH <sub>2</sub>	31.8, CH <sub>2</sub>
16	22.6, CH <sub>2</sub>	22.6, CH <sub>2</sub>	22.6, CH <sub>2</sub>	22.8, CH <sub>2</sub>
17	14.1, CH <sub>3</sub>	14.1, CH <sub>3</sub>	13.9, CH <sub>3</sub>	13.8, CH <sub>3</sub>

lute configuration of C-3 and C-9 chiral centers could not be determined by the Mosher method due to the scarce amount of sample. However, the 3S configuration was suggested analogous with co-occurring compounds **1–5** on the basis of the positive  $[\alpha]_D$  value, which was also observed for this member of *Eryngium* polyacetylenes.

*trans*-Epoxy-triquetrol (diastereomeric mixture **7a/7b**) exhibited a sodiated molecular peak at *m/z* 299.1629 in the HRESIMS spectrum, consistent with the molecular formula C<sub>17</sub>H<sub>24</sub>O<sub>3</sub>, with a difference of 16 mass units with respect to panaxydiol (**1**) and triquetridiol (**6**). The NMR spectra (see Tables 1 and 2) resembled those of the other co-occurring polyacetylenes, indicating the presence in the structure of the same terminal portions. The main difference was in the lack of the internal double bond being the olefinic resonances replaced by signals due to oxygenated carbons. However, a careful analysis of both <sup>1</sup>H and <sup>13</sup>C NMR spec-



**Fig. 3** Lignans and flavonoids isolated from *E. triquetrum*.

tra revealed that we were dealing with a mixture of two diastereoisomers (**7a/7b**). Unfortunately, every attempt to separate the two isomers failed, thus, they were characterized spectroscopically as a mixture. In the NMR spectra (recorded in CDCl<sub>3</sub> and C<sub>6</sub>D<sub>6</sub>), along with the expected series of signals due to terminal C-1/C-7 and C-11/C-17 fragments, resonances due to oxygenated methines were observed (Table 1 and 2). In particular, two sets of signals (ratio 1:0.8) were recognized in the proton and carbon spectra, and were connected by detailed analysis of <sup>1</sup>H-<sup>1</sup>H COSY and HSQC experiments. Two distinct spin systems accounting for the C-8/C-10 fragment of both isomers were defined, whereas the remaining portions of the molecules gave indistinguishable signals in the spectra as reported in Tables 1 and 2. For the main isomer (**7a**), the sequence was constituted by the proton resonating at δ<sub>H</sub> 4.66 (1H, app. t, J = 5.0 Hz, H-8), which had a cross-peak with the multiplet at δ<sub>H</sub> 3.02 (1H, m, H-9), which, in turn, correlated with the signal at δ<sub>H</sub> 3.10 (1H, dt, J = 2.0 and 5.6 Hz, H-10). An analogous sequence connecting sequentially H-8 (δ<sub>H</sub> 4.41, dd, J = 3.8 and 7.1 Hz), H-9 (δ<sub>H</sub> 3.00, m), and H-10 (δ<sub>H</sub> 2.98, dt, J = 2.0 and 5.6 Hz) was deduced for the minor component (**7b**). These data along with the <sup>13</sup>C NMR values [δ<sub>C</sub> **7a/7b** 61.3/62.3 (C-8), δ 58.8/59.3 (C-9), and δ 56.0/56.1 (C-10)] strongly suggested the presence of a hydroxyl group at C-8 and the epoxy ring, including C-9 and C-10. An NMR assignment was also made in C<sub>6</sub>D<sub>6</sub> (Table 1 and 2) due to the observation that the signals of the two isomers were better distinguished in this solvent. The relative configuration of the chiral centers C-9 and C-10 was deduced by comparing the proton and carbon value pattern for the hydroxy epoxide fragment in both **7a** and **7b** with literature NMR data for model compounds exhibiting a *cis* or *trans* epoxide [25–28]. In particular, the small coupling constants of epoxide protons (J<sub>H-H</sub> = 1–2 Hz) measured by homo-decoupling experiments, the NOE effects observed between H-8 and H-10, and the epoxide carbon values were consistent with a *trans* geometry, analogous to the synthetic C-10 epimer of oploxyne A

[28]. Having fixed the *trans* stereochemistry of the epoxide moiety, it was reasonable to suppose that the two isomers differed in the relative configuration of the carbinol carbon adjacent to the epoxide. Unfortunately, due to the very scarce amount of the sample, a further stereochemical investigation could not be made and, thus, this aspect remained undetermined. On the other side, the absolute configuration of C-3 was supposed to be the same as the co-occurring polyacetylenes by biogenetic considerations.

The lignan fraction was purified as reported in Materials and Methods to give an inseparable *erythro/threo* pair of unprecedented demethoxy carolignan Z (**8a/8b**) along with a series of known compounds. Previously reported lignans included two epoxy-neolignans, balanophonin (**9**) [29] and ficusal (**10**) [30], the diepoxy-lignan (+)-medioresinol (**11**) [31], four sesquiligans, buddlenol C (**12**) [32], buddlenol D (**13**) [32], and *threo*- and *erythro*-buddlenol E (**14**, **15**) [32, 33], and secolignan 9,9'-diferylolyl-secoisolariciresinol (**16**) [34] (Fig. 3).

Demethoxy carolignan Z was isolated and characterized as an *erythro/threo* pair (**8a/8b**). The molecular formula C<sub>40</sub>H<sub>40</sub>O<sub>13</sub> was deduced from the sodiated molecular peak at 751.2360 in the HRESIMS spectrum. <sup>1</sup>H and <sup>13</sup>C NMR data (Table 3) showed strong similarities with those reported in the literature for carolignans [35–37], suggesting the presence of a carolignan framework for **8a/8b**. The <sup>1</sup>H NMR spectrum (Table 3) showed a series of aromatic protons undistinguishable for the two isomers and attributable to two units of feruloyloxy group (A and B rings), two sets of signals for the 1,3,4-trisubstituted benzene units (C and D rings), two *trans* olefinic protons (H-7' and H-8'), and two methylene protons (H<sub>2</sub>-9'). Additional signals due to protons linked to oxygen-bearing carbons were also present in the spectrum (H-7, H-8, and H<sub>2</sub>-9). Careful analysis of these latter signals and their <sup>1</sup>H-<sup>1</sup>H COSY correlations clearly indicated that they constituted two distinct set of signals due to the 1,2,3-propanetriol moiety for each isomer. A spin system consisted in the proton at δ<sub>H</sub> 4.93 (1H, m, H-7) that had a cross-peak with the methine proton at δ<sub>H</sub> 4.30 (H-8), which was, in turn, coupled to the methylene protons at δ<sub>H</sub> 4.14 (1H, dd, J = 12.0, 3.6 Hz, H-9a) and δ<sub>H</sub> 4.32 (1H, dd, J = 12.0, 3.6 Hz, H-9b). In the other spin system, the proton at δ<sub>H</sub> 4.91 (1H, m, H-7) was coupled with the methine at δ<sub>H</sub> 4.52 (1H, m, H-8), which was correlated with the methylene protons at δ<sub>H</sub> 4.50 (1H, dd, J = 12.0, 3.6 Hz, H-9a) and δ<sub>H</sub> 4.32 (1H, dd, J = 12.0, 3.6 Hz, H-9b). The HSQC spectrum showed two sets of carbon values for the two sequences: δ<sub>C</sub> 72.3 (C-7), δ<sub>C</sub> 84.4, (C-8), and δ<sub>C</sub> 62.6 (C-9) for the *erythro*-isomer (**8a**) and δ<sub>C</sub> 74.4 (C-7), δ<sub>C</sub> 86.1, (C-8), and δ<sub>C</sub> 63.2 (C-9) for the *threo*-isomer (**8b**), according to literature data reported for *erythro/threo* carolignans isolated from *Ochroma lagopus* [38]. Diagnostic HMBC correlations aided us in linking all these moieties. The carbonyl C-9'' showed cross-peaks with H-8'' and H-9a connecting one of the two feruloyloxy groups (ring A) to the propanetriol moiety, whereas the second carbonyl C-9''' had long-range correlations with both H-8''' and H<sub>2</sub>-9', linking the second feruloyloxy group (ring B) to ring D. Phenolic carbons C-1 and C-4' were correlated with H-7 and 7-OH and with H-8 and H-2''/H-6'', respectively, thus linking both 1,3,4-trisubstituted rings C and D to the propanetriol moiety. By these data, the proposed structure for the *erythro/threo* pair **8a/8b** was diferuloyloxy-4,8-dihydroxy-3,3'-dimethoxy-4',8'-oxyneolignan-7'-en-9,9'-dioate, corresponding to the 5'-demethoxy derivative of carolignan Z, which has been very recently described from an *Euphorbia* species [37].

Finally, four known phenolic compounds were identified in the extract: *threo*-2,3-bis(4-hydroxy-3-methoxyphenyl)-3-butoxy-

propan-1-ol (**17**) [39], *erythro*-2,3-bis(4-hydroxy-3-methoxyphenyl)-3-butoxypropan-1-ol (**18**) [39], kaempferol-3-*O*-(2,6-di-*Z*-*p*-coumaroyl)-glucoside (**19**) [40], and kaempferol-3-*O*-(2,6-di-*E*-*p*-coumaroyl)-glucoside (**20**) [40] (● Fig. 3).

In conclusion, the chemical study on *E. triquetrum* from Algeria resulted in the isolation of 20 compounds, polyacetylenes, lignans, and flavonoids, including previously undescribed metabolites. It is the first report of polyacetylenes and lignans from this plant. The secondary metabolite pattern is, however, in agreement with the literature data for other *Eryngium* species.

## Materials and Methods

### General experimental procedures

Optical rotations were measured on a Jasco DIP 370 digital polarimeter. FTIR spectra were obtained using a Spectrum 100 instrument from Perkin Elmer fitted with a Germanium/KBr beam splitter and a deuterated tryglycine sulfate (DTGS) wideband detector on KBr pellets. UV spectra (MeOH) were acquired on a Jasco V-650 spectrophotometer. ESIMS were performed on a Micromass Q-TOF MicroTM coupled with an HPLC Waters Alliance 2695. The instrument was calibrated by using a PEG mixture from 200 to 1000 MW (resolution specification 5000 FWHM, deviation < 5 ppm RMS in the presence of a known lock mass). High-resolution mass spectra (HRESIMS) were acquired on a Q-Exact hybrid quadrupole-orbitrap mass spectrometer (Thermo Scientific). NMR experiments were recorded at the ICB-NMR Service Centre. Chemical shifts values are reported in ppm and referenced to the internal signals of residual protons (CDCl<sub>3</sub>, <sup>1</sup>H δ 7.26, <sup>13</sup>C 77.0 ppm; C<sub>6</sub>D<sub>6</sub>, <sup>1</sup>H δ 7.15, <sup>13</sup>C 128.0 ppm). 1D and 2D NMR spectra were acquired on a Bruker Avance-400 operating at 400 MHz using an inverse probe fitted with a gradient along the Z-axis and a Bruker DRX-600 operating at 600 MHz using an inverse TCI CryoProbe fitted with a gradient along the X-axis. HPLC separation was performed on a Shimadzu high-performance liquid chromatography using a Shimadzu liquid chromatograph LC-10AD equipped with an UV SPD-10A wavelength detector with a reversed-phase (RP) column (10 × 250 mm, Aventis-Supelco). Silica gel chromatography was performed using pre-coated Merck F254 plates (TLC) and Merck Kieselgel 60 powder (70–230 mesh). The AgNO<sub>3</sub> silica gel was prepared by adsorbing a silver nitrate solution on silica (10% w/w). The spots on TLC were visualized under UV light (254 nm) and then were sprayed with 10% H<sub>2</sub>SO<sub>4</sub> in water followed by heating.

### Plant material

The plant *E. triquetrum* was collected in April 2012 in Merouana (Batna, Algeria) and identified by Prof. Bachir Oudjehih, Institute of Agronomy of University of Batna (Algeria). A voucher specimen has been deposited in the herbarium of the department of the same University with code 122/ISVSA/UHL/13.

### Extraction and isolation

The dried and powdered aerial part (500 g) of *E. triquetrum* was exhaustively extracted at room temperature with a hydroalcoholic solution (MeOH/H<sub>2</sub>O 8 : 2) three times (400 mL×3), filtered, combined, and concentrated under vacuum to afford an aqueous solution. The latter was sequentially partitioned with light petroleum ether (250 mL×3), ethyl acetate (250 mL×3), and *n*-butanol (250 mL×3) to give 2.4 g of light petroleum ether, 3.0 g of ethyl acetate, and 11.0 g of butanolic extracts, respectively. The ex-

**Table 3** <sup>1</sup>H and <sup>13</sup>C NMR (400 MHz) data in CDCl<sub>3</sub> for **8a** (*erythro* isomer). Data in square brackets refer to **8b** (*threo* isomer).

1	–	131.2, C
2	7.04, d (2.0)	109.2, CH
3	–	145.2, C
4	–	146.8, C
5	6.85, d (8.5)	114.2, CH
6	6.88, dd (8.5, 2.0)	119.2, CH
7	4.93, m [4.91]	72.3, CH [74.4]
8	4.30, m [4.52]	84.4, CH [86.1]
9	4.14, dd (12.0, 3.6) [4.50, dd (12.5, 3.6)] 4.32, dd (12.0, 3.6) [4.32, dd (12.0, 3.6)]	62.6, CH [63.2]
3-OMe	3.86, s	55.8, CH <sub>3</sub>
1'	–	133.5, C
2'	6.95, d (1.5)	109.3, CH
3'	–	150.1, C
4'	–	148.1, C
5'	7.12, d (8.5)	120.2, CH
6'	7.01, dd (8.5, 1.5)	120.4, CH
7'	6.64, d (16.0)	133.6, CH
8'	6.26, dt (16.0, 6.5)	122.9, CH
9'	4.85, d (6.5)	65.1, CH <sub>2</sub>
3''-OMe	3.89, s	55.7, CH <sub>3</sub>
1''	–	126.3, C
2''	7.01, d (1.8)	110.0, CH
3''	–	145.3, C
4''	–	146.8, C
5''	6.91, d (8.3)	114.7, CH
6''	7.07, dd (8.3, 1.8)	123.7, CH
7''	7.49, d (16.0)	145.5, CH
8''	6.23, d (16.0)	114.9, CH
9''	–	166.4, C
3'''-OMe	3.93, s	55.7, CH <sub>3</sub>
1'''	–	126.7, C
2'''	7.01, d (1.8)	109.3, CH
3'''	–	145.5, C
4'''	–	147.2, C
5'''	6.94, d (8.3)	114.3, CH
6'''	7.08, dd (8.3, 1.8)	123.7, CH
7'''	7.65, d (16.0)	144.9, CH
8'''	6.33, d (16.0)	114.7, CH
9'''	–	166.8, C
3''''-OMe	3.92, s	55.8, CH <sub>3</sub>

<sup>a</sup> Assignments aided by COSY, HSQC, and HMBC

tracts were analyzed by TLC chromatography, which revealed an interesting metabolite pattern in the ethyl acetate extract. A portion of this extract (2.0 g) was fractionated by Sephadex LH-20 chromatography (column diameter: 3 cm, h: 130 cm, LH-20: 200 g) using a solution of CHCl<sub>3</sub>/MeOH (1 : 1) in isocratic fashion to get 50 fractions (each with an 8-mL volume) that were combined on the basis of their TLC chromatographic behavior to give 11 fractions (1–11), two of which (fractions 4 and 7) were considered in this work. Fraction 4 (580 mg) was subjected to silica gel column chromatography (column diameter: 2.5 cm, h: 100 cm silica gel: 29 g) eluted with an increasing polarity gradient of CHCl<sub>3</sub>/MeOH. A total of 70 fractions were collected (volume of each tube: 15 mL) starting from CHCl<sub>3</sub> (250 mL of solvent, collected 17 tubes), CHCl<sub>3</sub>/MeOH 98 : 2 (200 mL of solvent, 13 tubes), CHCl<sub>3</sub>/MeOH 95 : 5 (200 mL of solvent, 13 tubes), CHCl<sub>3</sub>/MeOH 9 : 1 (300 mL of solvent, 15 tubes), CHCl<sub>3</sub>/MeOH 7 : 3 (170 mL of solvent, 11 tubes), and only MeOH (150 mL, 10 tubes). After TLC chromatography, all of these fractions were combined

to give 12 final fractions (F4-1→F4-12), which were analyzed by  $^1\text{H}$  NMR. Subfraction F4-1 (10.0 mg) resulted in containing pure falcarinol **2** (Rf 0.90,  $\text{CHCl}_3/\text{MeOH}$  95:5, 8 mg). Subfraction F4-3 (60 mg) was a mixture that was further purified by HPLC (Supelco, Ascentis C18 column  $1.0 \times 25$  cm) with a 45-min linear gradient from 60 to 100% MeOH in  $\text{H}_2\text{O}$  (flow rate 2 mL/min) to yield pure compounds **1** ( $t_{\text{R}}$  34.8 min, 20 mg), **4** ( $t_{\text{R}}$  35.5 min, 3.0 mg), and **3** ( $t_{\text{R}}$  37.5 min, 2.0 mg), along with a mixture collected as single peak at  $t_{\text{R}}$  33.5 min. This mixture (19.3 mg) was subjected to a further HPLC purification (Supelco, Ascentis C18 column,  $0.46 \times 25$  cm) using a 40-min linear gradient starting from 70 to 100% MeOH in  $\text{H}_2\text{O}$  (flow rate 1 mL/min) to obtain pure compounds **5** ( $t_{\text{R}}$  15.8 min, 4 mg) and **1** ( $t_{\text{R}}$  17.2 min, 8 mg). Subfraction F4-4 (43.0 mg) was purified by HPLC using the same condition described for F4-3 (Supelco, Ascentis C18 column  $1.0 \times 25$  cm) yielding pure compounds **11** ( $t_{\text{R}}$  11.3 min, 4.0 mg), **17** ( $t_{\text{R}}$  23.5 min, 3.0 mg), **18** ( $t_{\text{R}}$  24.2 min, 4.0 mg), and **4** ( $t_{\text{R}}$  36.6 min, 2.0 mg), along with compound **6** collected in a mixture with **5** at  $t_{\text{R}}$  34 min. This mixture (3.0 mg) was purified on an  $\text{AgNO}_3$ -impregnated  $\text{SiO}_2$  pipette column ( $\text{AgNO}_3$  silica gel: 1 g, volume of each collected tube: 1 mL) eluted with a gradient of diethyl ether in hexane (hexane/diethyl ether 8:2, 10 mL; hexane/diethyl ether 7:3, 10 mL; hexane/diethyl ether 6:4, 10 mL; hexane/diethyl ether 5:5, 20 mL; hexane/diethyl ether 4:6, 10 mL) to afford pure compounds **5** (1.0 mg, eluted with hexane/diethyl ether 7:3) and **6** (0.8 mg, eluted with hexane/diethyl ether 6:4). Subfraction F4-5 (5.0 mg) was subjected to HPLC purification (Supelco, Ascentis C18 column,  $0.46 \times 25$  cm) using a 40-min linear gradient from 70 to 100% MeOH in  $\text{H}_2\text{O}$ , (flow rate 1 mL/min) obtaining the mixture **7a/7b** ( $t_{\text{R}}$  12.5 min, 1.0 mg). Subfraction F4-6 (100.0 mg) was purified by HPLC (Supelco, Ascentis C18 column,  $1.0 \times 25$  cm) with a 60-min linear gradient from 50 to 100% MeOH in  $\text{H}_2\text{O}$  (flow rate 2 mL/min) to afford pure compounds **10** ( $t_{\text{R}}$  22.1 min, 2.0 mg), **9** ( $t_{\text{R}}$  26.3 min, 7.2 mg), **13** ( $t_{\text{R}}$  27.1 min, 3.2 mg), **12** ( $t_{\text{R}}$  29.7 min, 6.4 mg), **14** ( $t_{\text{R}}$  31.1, 3.0 mg), and **16** ( $t_{\text{R}}$  45.0, 2.5 mg), together with two fractions collected at  $t_{\text{R}}$  28.5 (fraction A) and 43.2 min (fraction B), respectively. Both fractions were subjected to a further HPLC purification (Supelco, Ascentis C18 column  $1 \times 25$  cm) by using an isocratic elution ( $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ , 1:1 for fraction A;  $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ , 6:4 for fraction B) to get pure compound **15** ( $t_{\text{R}}$  8.3 min, 4.0 mg) from fraction A and the mixture **8a/8b** ( $t_{\text{R}}$  12.5 min, 3.5 mg) from fraction B. Fraction 7 (130 mg), from the starting Sephadex column, was further chromatographed on a silica gel column (column diameter: 1.5 cm; h: 60 cm; silica gel: 8 g) employing  $\text{CHCl}_3/\text{MeOH}$  in order of increasing polarity. A total of seven subfractions were collected (each fraction volume: 60 mL; subfraction 1 eluted with  $\text{CHCl}_3$ ; subfraction 2 eluted with  $\text{CHCl}_3/\text{MeOH}$  98:2; subfraction 3 eluted with  $\text{CHCl}_3/\text{MeOH}$  95:5; subfraction 4 eluted with  $\text{CHCl}_3/\text{MeOH}$  9:1; subfraction 5 eluted with  $\text{CHCl}_3/\text{MeOH}$  8:2; subfraction 6 eluted with  $\text{CHCl}_3/\text{MeOH}$  7:3; subfraction 7 eluted with MeOH). Subfraction 4 (13.5 mg), obtained from  $\text{CHCl}_3/\text{MeOH}$  9:1, was a mixture of *cis* and *trans* *p*-coumaroyl-kaempferolglucosides. This mixture was further separated by HPLC (Supelco, Ascentis C18 column,  $1.0 \times 25$  cm) by using an isocratic elution ( $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ , 1:1, flow rate 2.0 mL/min) to give pure compounds **20** ( $t_{\text{R}}$  9.8 min, 3.5 mg) and **19** ( $t_{\text{R}}$  10.5 min, 4.0 mg). (*Z*)-heptadeca-1,8-diene-4,6-diyne-3,10-diol (**5**), *cis*-panaxydiol: Oil;  $^1\text{H}$  and  $^{13}\text{C}$  NMR see **Tables 1** and **2**; HRESIMS  $m/z$  283.1670 (calcd. for  $\text{C}_{17}\text{H}_{24}\text{O}_2$  Na, 283.1674). (*10E*)-heptadeca-1,10-diene-4,6-diyne-3,9-diol (**6**), triquetridiol: Oil;  $[\alpha]_{\text{D}}^{20} + 35.42$  (c 0.5,  $\text{CHCl}_3$ ); UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ): 257

(1.98), 243 (2.07), 230 (2.3), 203 (3.18); IR  $\nu_{\text{max}}$ : 3351, 2955, 2925, 2853, 2255, 1731, 1025, 930, 705  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR see **Tables 1** and **2**; HRESIMS  $m/z$  283.1665 (calcd. for  $\text{C}_{17}\text{H}_{24}\text{O}_2\text{Na}$ , 283.1674).

Heptadeca-1-ene-9,10-epoxy-4,6-diyne-3,8-diol (**7a/7b**), triquetrol: Oil; UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ): 284 (2.75), 269 (2.82), 257 (2.90), 244 (2.95), 232 (2.97); IR  $\nu_{\text{max}}$  3382, 2856, 2255, 2154, 1721, 1463, 1024, 930  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR in  $\text{CDCl}_3$  see **Tables 1** and **2**; HRESIMS  $m/z$  299.1629 (calcd. for  $\text{C}_{17}\text{H}_{24}\text{O}_3$  Na, 299.1623).

Diferuloyloxy-4,8-dihydroxy-3,3'-dimethoxy-9,9'-4,8-oxyneli-gnan-7'-en-9,9'-di-oate (**8a/8b**), 5'-demethoxy carolignan Z: White powder; UV (MeOH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ): 324 (4.017), 288 (3.93), 233 (4.019), 219 (4.12), 204 (4.30); IR  $\nu_{\text{max}}$  3410, 3011, 2956, 2927, 2852, 1701, 1630, 1594, 1512, 1430, 1377, 1268, 978, 819, 755  $\text{cm}^{-1}$ ;  $^1\text{H}$  and  $^{13}\text{C}$  NMR see **Table 3**; HRESIMS  $m/z$  751.2360 (calcd. for  $\text{C}_{40}\text{H}_{40}\text{O}_{13}$  Na, 751.2367).

(8E)-heptadeca-1,8-diene-4,6-diyne-3S,10-diol (**1**), *trans*-panaxydiol: Oil;  $^1\text{H}$  NMR values in ppm ( $\text{CDCl}_3$ , 400 MHz):  $\delta_{\text{H}}$  6.33 (1H, dd,  $J = 15.9, 6.0$  Hz, H-9), 5.95 (1H, ddd,  $J = 15.9, 10.3, 5.4$  Hz, H-2), 5.77 (1H, d,  $J = 15.9$  Hz, H-8), 5.47 (1H, d,  $J = 16.5$  Hz, H-1a), 5.25 (1H, d,  $J = 10.3$  Hz, H-1b), 4.98 (1H, app t,  $J = 5.4$  Hz, H-3), 4.18 (1H, m, H-10), 1.53 (2H, m, H<sub>2</sub>-11), 1.33 (2H, m, H<sub>2</sub>-12), 1.29 (2H, m, H<sub>2</sub>-13), 1.29 (2H, m, H<sub>2</sub>-14), 1.27 (2H, overlapped, H<sub>2</sub>-15), 1.27 (2H, overlapped, H<sub>2</sub>-16), 0.88 (3H, t,  $J = 7.1$  Hz, H<sub>3</sub>-17);  $^{13}\text{C}$  NMR values in ppm ( $\text{CDCl}_3$ , 75 MHz):  $\delta_{\text{C}}$  149.8 (CH, C-9), 136.0 (CH, C-2), 117.0 (CH<sub>2</sub>, C-1), 108.1 (CH, C-8), 80.0 (C, C-4), 77.6 (C, C-7), 73.6 (C, C-6), 72.0 (CH, C-10), 70.9 (C, C-5), 63.5 (CH, C-3), 36.9 (CH<sub>2</sub>, C-11), 31.7 (CH<sub>2</sub>, C-15), 29.4 (CH<sub>2</sub>, C-13) 29.2 (CH<sub>2</sub>, C-14), 25.2 (CH<sub>2</sub>, C-12), 22.6 (CH<sub>2</sub>, C-16), 14.1 (CH<sub>3</sub>, C-17).

#### Preparation of MTPA esters of compound 1

(*R*)- and (*S*)-MTPA-Cl (10  $\mu\text{L}$ ) and a catalytic amount of DMAP were separately added to two different aliquots of panaxydiol (**1**) (1.0 mg each) in dry  $\text{CH}_2\text{Cl}_2$  (0.5 mL). The resulting mixtures were allowed to stand at rt for 12 h. After evaporation of the solvent, the mixtures were purified on a  $\text{SiO}_2$  pipette Pasteur ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$ , 99:1) affording pure (*S*)- and (*R*)-MTPA esters of **1**, respectively.

(*S*)-MTPA ester of **1**: selected  $^1\text{H}$  NMR values ( $\text{CDCl}_3$ , 600 MHz):  $\delta_{\text{H}}$  5.529 (1H, d,  $J = 17.0$  Hz, H-1a), 5.362 (1H, d,  $J = 10.2$  Hz, H-1b), 5.834 (1H, ddd,  $J = 5.5, 10.2, 17.0$  Hz, H-2), 6.153 (1H, d,  $J = 5.7$  Hz, H-3); ESIMS  $m/z$  715 [M + Na]<sup>+</sup>.

(*R*)-MTPA ester of **1**: selected  $^1\text{H}$  NMR values ( $\text{CDCl}_3$ , 600 MHz):  $\delta_{\text{H}}$  5.611 (1H, d,  $J = 17.0$  Hz, H-1a), 5.422 (1H, d,  $J = 10.2$  Hz, H-1b), 5.939 (1H, ddd,  $J = 5.5, 10.2, 17.0$  Hz, H-2), 6.126 (1H, d,  $J = 5.7$  Hz, H-3); ESIMS  $m/z$  715 [M + Na]<sup>+</sup>.

(3S,9Z)-Heptadeca-1,9-diene-4,6-diyne-3-ol (**2**), falcarinol: Oil;  $^1\text{H}$  NMR values in ppm ( $\text{CDCl}_3$ , 400 MHz):  $\delta_{\text{H}}$  5.94 (1H, ddd,  $J = 15.7, 10.1, 5.4$  Hz, H-2), 5.51 (1H, m, H-10), 5.46 (1H, d,  $J = 15.7$  Hz, H-1a), 5.37 (1H, m, H-9), 5.24 (1H, d,  $J = 10.1$  Hz, H-1b), 4.91 (1H, app t,  $J = 5.4$  Hz, H-3), 3.03 (2H, d,  $J = 7.0$  Hz, H<sub>2</sub>-8), 2.02 (2H, m, H<sub>2</sub>-11), 1.37 (2H, m, H<sub>2</sub>-12), 1.27 (2H, overlapped, H<sub>2</sub>-13), 1.29 (2H, m, H<sub>2</sub>-14), 1.27 (2H, overlapped, H<sub>2</sub>-15), 1.25 (2H, overlapped, H<sub>2</sub>-16), 0.88 (3H, t,  $J = 7.1$  Hz, H<sub>3</sub>-17);  $^{13}\text{C}$  NMR values in ppm ( $\text{CDCl}_3$ , 75 MHz):  $\delta_{\text{C}}$  136.2 (CH, C-2), 133.0 (CH, C-10), 116.9 (CH<sub>2</sub>, C-1), 122.1 (CH, C-9), 80.0 (C, C-7), 74.5 (C, C-4), 71.1 (C, C-5), 64.3 (C, C-6), 63.3 (CH, C-3), 31.9 (CH<sub>2</sub>, C-15), 29.3 (CH<sub>2</sub>, C-13) 29.3 (CH<sub>2</sub>, C-14), 27.2 (CH<sub>2</sub>, C-11), 29.2 (CH<sub>2</sub>, C-12), 22.7 (CH<sub>2</sub>, C-16), 17.7 (CH<sub>2</sub>, C-8), 14.1 (CH<sub>3</sub>, C-17).





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